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Reaction of Hexafluoroacetone with Aminophosphites Containing the 2,2,2-Trifluoro-1-(trifluoromethyl)ethyl Grouping

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The reactions of dihaloaminophosphines RNHPF $_2$ (R=H, Me, tBu) and R $_2$ NPCl $_2$ (R=Me, Et, SiMe $_3$; R $_2$ =CH $_2$ (CH $_2$ CMe $_2$) $_2$ with LiOCH(CF $_3$) $_2$ yield the corresponding aminophosphites R $_2$ NP[OCH(CF $_3$) $_2$] $_2$. Hexafluoroacetone reacts with RNH[OCH(CF $_3$) $_2$] $_2$ as well as MeNHPF $_2$ and tBuNHPF $_2$ in good yields to the 1,3,5 λ^5 -oxazaphosphetanes $\underline{1}$ - $\underline{4}$, which show rapid pseudorotation at room temperature.

RNHPX,
$$2(CF_3)_2CO$$
 \longrightarrow $\begin{pmatrix} (CF_3)_2C-O & \underline{1}: R=H, X=OCH(CF_3)_2; \underline{2}: R=CH_3, X=F, \\ N-P & X & \underline{3}: R=CH_3, X=OCH(CF_3)_2; \underline{4}: R=tBu, X=F \end{pmatrix}$

The X-ray structure of $\underline{1}$ exhibits a distorted trigonal bipyramid. With dialkylaminophosphites hexafluoroacetone gives the monocyclic phosphoranes $\underline{5}$ and $\underline{6}$, whereas with aminophosphites containing bulky alkylgroups no reaction occurs. A replacement of the $(CF_3)_2CHO$ -groups by smaller substituents results in easier addition of hexafluoroacetone. Therefore the aminophosphines tBuP(X)NEt $_2$ (X=F, C1, OCH(CF $_3$) $_2$) react with hexafluoroacetone in high yields to the 1,3,2 λ^5 -dioxaphospholanes $\underline{7}$ - $\underline{9}$. The phosphoranes $\underline{5}$ - $\underline{9}$ show no ligand exchange process at room temperature.